

http://www.lccp.polimi.it

Dipartimento di Energia – Politecnico di Milano







- Chemical Engineering and Catalysis
- Catalysis @ PoliMI
- Doing your thesis @ LCCP: research activities and open positions



- Catalysis the process by which the rate and products of chemical reactions are altered by substance unchanged by the reaction – it is at the core of the chemical and petroleum industries.
- Catalysis-based processes represent more than 90% of current chemical processes and generate 60% of today's chemical products.
- In addition to traditional roles in the chemical and petroleum industries, catalysts are of growing importance in fields ranging from environmental protection and energy to pharmaceuticals and the processing of high performance materials.



"The urgent need for fuels in an era of declining resources and of pressing environmental concerns demands a resurgence in catalysis science (and engineering)» Report from the U.S. DOE



Catalytic processes for energy





Laboratory of Catalysis and Catalytic Processes

- ✓ The Laboratory of Catalysis and Catalytic Processes (LCCP) is a research group at Politecnico di Milano that has pioneered multidisciplinary research in the science and engineering of catalysis.
- ✓ Since in the next years catalysis is expected to play a crucial role in energy-related fields, the LCCP has recently moved in the Department of the Energy of Politecnico di Milano, keeping though its strong connection with the School of Chemical Engineering.
- ✓ The mission of the LCCP is the education of students via relevant research in the multidisciplinary field

of catalysis science, spanning from fundamental research to technical applications, with a strong connection to the industrial world.







Permanent staff

Pio Forzatti (full professor) Enrico Tronconi (full professor) Luca Lietti (full professor) Gianpiero Groppi (full professor) Alessandra Beretta (full professor) Isabella Nova (associate professor) Lidia Castoldi (assistant professor) Alessandro Donazzi (assistant professor) Matteo Maestri (assistant professor)

Carlo Giorgio Visconti (assistant professor) Roberto Matarrese (assistant professor)

PhD Students

Fabio Marchitti Amir Reza Fahami Laura Fratalocchi Luca Dietz Stefano Rebughini Andrea Carrera Nicola Usberti Pedram Aghei Tommaso Selleri

Mauro Bracconi Leonardo Falbo Morteza Rahmanipour

Post-Doc Researchers

Rossella Bonzi Michela Martinelli Laura Righini Maria Pia Ruggeri



Senior Scientist Natale Ferlazzo

Temporary researchers 50 undergrads-masters/year

Visiting scientists from abroad 5-10 people/year

Laboratory of Catalysis and Catalytic Processes

Technicians

Roberto Losi

Enrico Aliprandi

Enrica Ceresoli

Academic collaborations



France: Sweden: **University of Caen** Chalmers University of Technology **KTH Royal Institute of Technology** Germany: Fritz-Haber-Institut der MPG, Berlin **TUM**, Munich **DLR**, Stoccarda ✓ KIT, Karlsruhe Czech Republic: Academy of Sciences, Prague Italy: ✓ CNR Politecnico di Torino Università dell'Aquila USA: \checkmark Spain: Università di Bologna ✓ Lehigh University ✓ University of Malaga ✓ Università di Genova University of California at Berkeley \checkmark **University of Madrid** Università "La Sapienza" di Roma **University of Delaware University of Siviglia** Università di Salerno University of Minnesota Università di Torino **Oak Ridge National Laboratory** ✓ Università di Udine **University of Houston** Università di Cagliari University of Kentucky (CAER)

Industrial collaborations









Several research activities are on-going at LCCP, involving from undergraduate to PhD students, coordinated by professional researchers.

Undergraduate/Bachelor thesis:

A short-term project introduces the student to the methodologies of research.

Master Thesis:

A mid-term project brings the student within a research team and a research field; original results and incremental advancement of research are the expected outcomes.

Ph.D. Thesis:

A long-term project is individually carried out by the student, who also completes his engineering education. This experience is expected to form highly qualified chemical engineers, able to "produce" innovation.



Since catalysis is a typical cross-multidisciplinary science, available theses involve activities in the following areas:

- Preparation and characterization of novel materials
- ✓ Reactivity studies: experimental investigations
- ✓ Development of kinetic schemes
- ✓ Modelling of chemical reactors

→ Experimental <u>and</u> modeling skills are developed!







Laboratory of Catalysis and Catalytic Processes

New LCCP laboratories at Campus Bovisa (B18):

- ✓ 3000 m²
- Laboratories for catalyst preparation and characterization, for catalytic testing under atmospheric and high-pressure







Equipment for catalyst preparation



Powdered catalysts:

Dry impregnation, Wet impregnation, Co-precipitation

Coated items:

Ball-milling, Ultracentrifugation Unit, deep coating, spraying, Rheometer







<u>Bulk monoliths</u>: Mixer, Kneader, Screw-Extruder,

Standard techniques: Climatic Chamber, Filtering Equipments, pH-meters, Drying and Calcination Ovens





Equipment for catalyst characterisation (morphological, structural, bulk and surface physico-chemical) GC-MS, TPD/R/O, FTIR, UV-Vis, XRD, DTA-TG, SEM available inside the group/ Department







In situ FTIR, probe and labelled molecule spectroscopies, HRTEM, XPS through collaboration with research groups with specific expertise outside our University





20 rigs for catalyst testing

(powder and structured catalysts, steady state and transient conditions, also operated under pressure)

Computational facilities

(computing time at supercomputer centers, software for modelling and simulation)











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Department of Energy

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Energy conversion:

- Catalytic Partial Oxidation of NG/LPG to CO/H₂
- Fischer-Tropsch Synthesis
- CO₂ activation

Environmental protection:

- Catalytic Combustion of Methane Emissions
- NO_x Storage-Reduction in Vehicles
- Combined Soot Combustion and NO_x Removal
- NH₃-SCR of NO_x for mobile sources
- NH₃-SCR of NO_x for stationary sources

Advanced reactor design and modelling:

- Novel Structured Catalytic Reactors
- First-principles Guided Chemical Engineering

Electrocatalysis:

- Water Splitting
- Solid Oxide Fuel Cells





Energy conversion

Catalytic Partial Oxidation of HC and bio-fuels to CO/H2





Syngas Platform Thermochemical



Energy efficiency Improved combustion through H2 injection (Power Plants/GT/ICE)

Ener Applicatic combinec system auxiliar

6





Novel applications of syngas call for <u>easily scalable</u> and <u>thermal efficient</u> syngas production in alternative to the traditional steam reforming technology

Portable Reformer, Time = 10⁻³ s





KINETICS AND MECHANISM

We study the reaction paths through which hydrocarbon fuels (NG, LPG, gasoline, diesel, ethanol...) convert into H_2 -rich streams over Rh nanoparticles

REACTOR DESIGN

We study the autothermal partial oxidation of the HC fuels at the pilot scale, with focus on the thermal behavior

TOOLS

We have developed and are developing advanced tools both for the experimental investigation (in situ sampling technique) and the modeling investigation (microkinetic and reactor modelling)

EXCHANGE PROGRAMS: NTNU, KIT



• Catalytic process for the conversion of natural gas, coal or biomasses into highquality diesel fuels and chemicals (e.g. Shell V-power diesel)

$$CO + 2 H_2 \longrightarrow C_n H_{2n+2}$$

$$C_n H_{2n} + H_2 O$$

$$C_n H_{2n+1} OH$$

• Main achievements in the last years (in cooperation with eni)

Development of lumped and detailed **kinetic models**, now used for the simulation of a pilot-scale demonstrative reactor (Sannazzar de' Burgundi eni's refinery) and the design of industrial reactor units

Development of an innovative compact reactor technology, based on structured catalysts (WO2010/130399 & WO2014/102350) successfully tested at the pilot scale (eni labs in San Donato)





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Laboratory of Catalysis and Catalytic Processes

Main achievements





Experimental activities:

- Preparation and characterization of innovative catalysts;
- Activity testing on a lab-scale fixed bed reactor running 24/7;

Modeling activities:

Modeling of innovative reactors, loaded with structured catalysts. Reactor optimization and experiment design;

Process analysis:

 Identification of a convenient technology to exploit Sulcis (Sardinia, Italy) coal reserves



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· 240 239

238 237

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- CO₂ to methane (power to gas)
- CO₂ (+CO) hydrogenation to fuels



CO₂ (+CO) hydrogenation to chemicals

✓ light olefins



Maire

ecnimont





Experimental activities:

Open positions

- Preparation and characterization of innovative catalysts;
- Activity testing on a lab-scale fixed bed reactor running 24/7;

Modeling activities:

- Kinetic modeling;
- Modeling of innovative reactors;
- Reactor optimization and experiment design;

Process analysis:

 Identification of a convenient technology to exploit each CO₂ sources











Environmental protection



1- Aftertreatment: abatement of CH₄ (a powerful greenhouse gas) emissions from NGV



2- Abatement of CH₄ emissions from stationary sources: e.g. geothermal process



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Studies on CH₄ oxidation performances under lean-rich switching conditions







NO_x storage-reduction (NSR) catalysts



S. Matsumoto, Cattech, 4 (2000) 2

- Investigated since 1999 within the framework of research project funded by the Italian Government (PRIN)
- Co-operation with Pirelli EcoTechnology (retrofit of Euro 4 vehicles) (2009-2011)
- Mechanistic study of the NO_x storage and reduction over LNT systems
 - Development of new catalytic materials
 - \checkmark Pathways for NO_x storage
 - ✓ Pathways for NO_x reduction with various reductants (H₂, CO, HCs)
 - Development of kinetic models for LNT catalysts operations (storage, reduction)



NO_x storage-reduction (NSR) catalysts



 understanding of the pathways involved in the NO_x storage and reduction (mechanistic studies with labeled molecules)

✓ modeling of experiments under operando condition – new spectrokinetic approach

in cooperation with University of Torino (IT), Goteborg (SE) and Caen (FR)



Diesel Particulate NOx Reduction (DPNR) catalysts



A catalytic wall-flow filter coated with a LNT catalyst is used to accomplish the simultaneous removal of soot and NO_x



- Investigated since 2007 within the framework of research project funded by the Italian Government (PRIN)
- Clarification of the interplay between De-NO_x activity and soot combustion activity of traditional NSR catalysts (Pt-Ba/Al₂O₃, Pt-K/Al₂O₃) and new formulations
- Study of the role of NO₂, NO+O₂, nitrate species and of alkaline and alkaline earth metal oxides in soot combustion







ongoing collaborations with

DAIMLER



• Participation in the FP7 EU project CO₂RE "CO₂ REduction for long haul transport"



•Participation in the EU H2020 project "Heavy Duty Gas Engines integrated into Vehicles – HDGAS (2015 – 2018)

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Since 2001 collaboration with

DAIMLER

 Transient 2D model of monolithic SCR converters used to design Euro 4, 5 and 6 compliant Mercedes-Benz Diesel vehicles commercialized in Europe and USA since 2005





Johnson Matthey

JM 🐼



 Just published Isabella Nova Enrico Tronconi *Editors*

Urea-SCR Technology for deNOx After Treatment of Diesel Exhausts





NH₃-SCR of NO_x

DAIMLER

Experimental and modeling activities at 3 different scales: microreactor (PoliMI), core monolith (PoliMI+Daimler), test bench (Daimler)







NH₃-SCR of NO_x

DAIMLER

- 1. Over new generation Cu-zeolites in <u>powder</u> form:
- ✓ Mechanistic study of the NO/NO₂ NH₃ SCR reactions
- ✓ Intrinsic kinetic analysis of the SCR reactions
- Deactivation effects
- 2. Over Cu-zeolite catalysts in <u>structured</u> form (honeycombs, foams ...):
- SCR reactivity study (role of mass transfer)
- ✓ Dynamic modeling of the structured SCR reactor





- EU project "HDGAS" (2015)
- Actual challenge:

Euro 6 regulations and low T activity!!





New generation of zeolite-based catalysts for SCR reactions





Experimental and modelling analysis of diffusional limitations in zeolite-based NH₃-SCR catalysts POLITECNICO DI MILANO

JMCX

Johnson Matthey





Combined Soot Combustion and NOX Removal - SDPF systems

New generation of catalytic filters for combined removal of soot and NOx: honeycomb wall flow monoliths washcoated with SCR catalyst

Experimental and modeling analysis of catalysts active in the combined NH₃-SCR of NOx and soot removal



Chart1



Analysis of byproducts formation on SCR catalysts



Experimental and modeling analysis of ammonium nitrate and N₂O formation mechanisms on different SCR catalysts





Although a well established technology, new issues are opened by the perspective of more stringent or new regulations with a need for further research and development.





Research lines:

1. Development of SCR catalysts with improved performances and minimum NH_3 slip even at high NH_3/NO inlet ratios (Target: increase plant efficiency of NOx reduction).

2. Modelling analysis of the combined NOx reduction and NH₃ oxidation.

- 3. Kinetic study and modelling of Hg° oxidation in SCR reactors.
- 4. Ageing of SCR catalysts: experimental strudy of the effect of ageing on the catalytic properties







Advanced reactor design and modelling



Novel Structured Catalytic Reactors

Open-cell foams as novel catalyst supports



➔ process intensification

e.g.: H₂ production in compact steam reformers







Novel Structured Catalytic Reactors

Open-cell foams as novel catalyst supports

- 1) Characterization and modeling of the geometry
- 2) Experimental investigation of mass and heat transfer properties
- 3) Development of generalized correlations for gas/solid mass and heat transfer



4) Preparation and testing of catalytic materials supported on foams





Novel Structured Catalytic Reactors

Conductive honeycombs and foams as novel catalyst supports:



for MeOH synthesis



TOTAL

- for CH₄ steam reforming
- for Fischer-Tropsch
- for methanation



Goal = intensification of radial heat transfer in tubular reactors







DEVELOPMENT

C N

optim

and

EMON

Concept applied to small scale MT-FBR for FTS



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First-principles guided chemical reaction engineering





First-principles guided chemical reaction engineering

MACROSCALE: COMPUTATIONAL FLUID DYNAMICS OF CATALYTIC REACTORS

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ELECTRONIC STRUCTURE CALCULATIONS AND MICROKINETIC MODELING









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- Electronic structure theory calculations: the finger-print of catalyst activity
- ✓ Validation and assessment of semiempirical methods
- ✓ Microkinetic model development
- ✓ Analysis of reaction mechanisms

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Available topics:

- Microkinetic model development and analysis of complex catalytic systems e.g., hydrocarbons activation, CO₂ activation
- First-principles assessment of catalysis by confinement in microporous siliceous zeolites (in collaboration with UC Berkeley, USA)
- ✓ First-principles studies of elementary steps at the catalyst surface (in collaboration with TUM, Munich, Germany and SISSA, Trieste)
- Advanced reactor design and optimization (in collaboration with MPI-Magdeburg, Germany)
- Advanced simulation of industrial catalytic reactors (in collaboration with BASF, Germany)
- ✓ First-principles analysis of transport properties in fixed bed reactors and foams (in collaboration with LCCP-"structured reactors")





Electrocatalysis



Hydrogen production via photoelectrolysis of water



Solid Oxide Fuel Cells (SOFCs)





SOFCs are the most efficient devices for the direct conversion of a fuel into electric power.

On-going research and open positions include:

- Development of anodic materials active and stable in the presence of hydrocarbon fuels and synthesis gas (low coke formation and low thermal deactivation).
- Investigation of the kinetic reaction pathway, the mass transport effects and the electrical effects of the different components of the SOFC.





Project

experimental modelling

CH ₄ geothermal	(Groppi/Forzatti)	Х	Х
$CH_4 NGV$	(Groppi/Forzatti)	Х	
CPO-mechanism	(Beretta/Donazzi/Groppi/Forzatti)	Х	Х
CPO-reactor design	"	Х	Х
Fischer-Tropsch	(Visconti/Lietti/Tronconi/Groppi)	Х	Х
CO_2 activation	(Visconti/Lietti/Forzatti)	Х	Х
LNT	(Castoldi//Lietti/Forzatti)	Х	Х
DPNR	"	Х	
SCR Daimler, JM, Cummins (Nova/Tronconi)		Х	Х
SCR MTU, HDGAS	(Nova/Tronconi)	Х	Х
SCR stationary	(Beretta/Forzatti)	Х	Х
Structured reactors	(Groppi/Tronconi/Visconti)	Х	Х
First-principle microkinetics (Maestri/Tronconi)			Х
H ₂ O splitting	(Matarrese/Nova)	Х	
SOFC	(Donazzi/Groppi)	Х	Х
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